Yunliang Yu,[‡] Changsheng Kuai,[‡] Remi Chauvin,^{*,‡,†}[®] Nian Tian,[‡] Shuangshuang Ma,[‡] and Xiuling Cui^{*,‡}[®]

[‡]Engineering Research Center of Molecular Medicine of Ministry of Education, Key Laboratory of Fujian Molecular Medicine, Key Laboratory of Xiamen Marine and Gene Drugs, School of Biomedical Sciences, Huaqiao University, Xiamen 361021, P. R. China

[†]CNRS, LCC (Laboratoire de Chimie de Coordination), 205, route de Narbonne, BP 44099, F-31077 Toulouse, Cedex 4 France and Université de Toulouse, UPS, INPT, F-31077 Toulouse, Cedex 4, France

S Supporting Information



ABSTRACT: The Rh-catalyzed *ortho*-C–H carbenoid insertion reaction of diarylazines with diazo compounds has been developed. A wide range of *ortho*-substituted diarylazines have been obtained in moderate to high yields with high regioselectivity at room temperature. The hydrolysis of the products could release ketones or aldehydes, giving access to aromatic 1,5-keto-diesters as valuable synthons for further chemical transformations.

INTRODUCTION

The transition-metal-catalyzed functionalization of C-H bonds has seen a tremendous development over the past decades. In particular, great progress has been made in site-selective chemical transformations of less reactive C-H bonds.² The usage of directing groups is an efficient way to improve reactivity and selectivity.³ Nitrogen-containing functional groups, such as pyridine,⁴ pyrimidine,⁵ imines,⁶ oximes,⁷ and hydrazones,⁸ have been widely explored as directing groups because of their coordination ability with transition-metal atoms. Azines (Ar(R)-C=N-N=C(R)Ar, which possess valuable intrinsic physical and biological properties⁹ and are useful organic synthetic intermediates,¹⁰ have recently proved to be a viable directing group by Huang¹¹ and Zhu (Scheme 1a).¹² Nevertheless, the formation of difunctionalized products of these nitrogen-containing substructures is difficult to avoid.¹³ Herein, the azine function is envisaged as a directing group for the Rh-catalyzed ortho-C-H carbenoid insertion reaction of diarylazines (Scheme 1b). In this protocol, the azine also acts as a bridging unit between two potentially reacting aromatic rings (Ar). Hereafter, two equivalent α -ketoaryl malonates could thus be obtained with diazomalonate in one step with a superior ortho-C-H site selectivity. The reaction could proceed smoothly at room temperature with broad functional group compatibility.

RESULTS AND DISCUSSION

Acetophenazine 1a and dimethyl diazomalonate 2a were first selected as model reactants to optimize the reaction conditions

(Table 1). The expected C-H insertion product 3aa was obtained in 89% isolated yield in the presence of $[Cp*RhCl_2]_2$ (2.5 mol %), AgSbF₆ (10 mol %), and PhCO₂H (25 mol %) in methanol at 25 °C in 12 h (entry 1). The structure of 3aa was confirmed by a single crystal X-ray diffraction analysis (Supporting Information, Figure S1). No product was observed in the absence of the Rh(III) catalyst (entry 2). The absence of Ag(I) salt resulted in a lower yield of 69% (entry 3). Only 26% yield of the desired product was obtained in the absence of acid (entry 4). Next, the use of alternative solvents, such as water, 1,4-dioxane, or 2,2,2-trifluoroethanol (TFE), was shown to provide inferior results (entries 5-7). Other Ag(I) salts were then investigated. Slightly lower yields were obtained with AgNO₃ or AgBF₄. A yield of 88% was restored with AgOAc (Table 1, entries 8-10 vs 1). The cheaper AgOAc salt was chosen as an additive instead of AgSbF₆. No reaction was observed in the presence of strong acids, such as HCl or trifluoroacetic acid (TFA) (entries 11 and 12), but the addition of AcOH led to a 81% yield (entry 13). Next, we examined the effect of benzoic acid additives and found that the combination of both AgOAc and 2,4,6-trimethylbenzoic acid as additives increased the yield to 97% (entries 14 and 15).

With optimized conditions in hand, the scope of arylketazine substrates was investigated with dimethyl diazomalonate 2a (Scheme 2). Aryl rings substituted with electron-donating groups were found to favor the carbenoid insertion of the

 Received:
 June 14, 2017

 Published:
 July 19, 2017

Article

pubs.acs.org/joc

Scheme 1. Azine-Directing-Group-Assisted ortho-C-H Functionalization



Table 1. Screening of Various Parameters for the Reaction ofAcetophenazine with Dimethyl Diazomalonate

	N-N	$ + 2 $ $ 0 = $ $ N_2 $ $ 0 = $ $ 0 = $ $ 0 = $ $ 0 = $ $ 0 = $ $ 0 = $ $ 0 = $ $ 0 = $ $ 0 = $	[Cp*RhCl ₂] ₂ (2.5 m Ag(I) acid, solven (1.5 r 12 h, rt, air	mL)	N N
	1a	2a		R= 0	CH(COOMe) ₂ 3aa
entry	Ag(I)	acid		solvent	yield ^{b} (%)
1	AgSbF ₆	PhCOOH	Ν	МеОН	89
2 ^c	AgSbF ₆	PhCOOH	Ν	MeOH	no
3	none	PhCOOH	Ν	MeOH	69
4	AgSbF ₆	none	Ν	MeOH	26
5	AgSbF ₆	PhCOOH	I	H ₂ O	19
6	AgSbF ₆	PhCOOH	1	,4-dioxane	trace
7	AgSbF ₆	PhCOOH	1	ГFE	no
8	$AgNO_3$	PhCOOH	Ν	МеОН	74
9	$AgBF_4$	PhCOOH	Ν	МеОН	81
10	AgOAc	PhCOOH	Ν	MeOH	88
11	AgOAc	HCl	Ν	MeOH	no
12	AgOAc	TFA	Ν	MeOH	no
13	AgOAc	AcOH	Ν	MeOH	81
14	AgOAc	p-nitrobenzoic a	ncid N	MeOH	87
15	AgOAc	2,4,6-trimethylb	enzoic acid N	МеОН	97

^aReaction conditions: 1a (0.2 mmol), 2a (0.45 mmol), $[Cp*RhCl_2]_2$ (2.5 mol %), Ag(I) (10 mol %), acid (25 mol %), solvent (1.5 mL), 12 h, under air, rt. ^bIsolated yields. ^cWithout $[Cp*RhCl_2]_2$. TFE: 2,2,2-trifluoroethanol. TFA: trifluoroacetic acid.

ortho-C-H bond, providing the dimalonyl products in high yields (3aa-3ga). para-Substituted acetophenazines afforded the anticipated products (3aa-3ea) in substantial yields (74-93%). The meta-methyl acetophenazine 1f gave the C-6-substituted product 3fa at the less hindered position in 94% yield. However, when a methoxy group or fluoro was placed at the meta position in 1g or 1h, the major products were the hindered C-2-substituted products, isolated with 63% (3ga) and 68%

yields (3ga), respectively, revealing an ortho-directing effect of the methoxy group and F atom. The reaction was found less efficient for ketazines substituted by electron-withdrawing groups (e.g., $-CF_{31}$, $-CN_{1}$, $-NO_{2}$) or halogens (Cl, Br, I). The meta-substitution with $-CF_3$, -CN, or $-NO_2$ did not afford the anticipated products, and the para-substitution by halogen atoms (Cl, Br) led to the monosubstituted product in low yields (<30%). When R² was ethyl, the targeted product (3ia) could be obtained in 85% yield. The other aryl-ketazine substrates, such as the α -tetralone azine (3j) and naphthyl methyl ketazine (3k), could give the corresponding products 3ja and 3ka both in 74% yields, albeit at 60 °C. Heating to 60 °C is also required for aldazine substrates, conceivably because of their lower coordinating ability as compared to that of ketazines. o-Methyl benzaldazine 11 provided the dimalonyl product 31a in 95% yield. m-Methyl benzaldazine 1m led mainly to 3ma, the dimalonylated product at the less hindered *ortho* position, in 75% yield (Scheme 2).

The scope of the diazo reactant was also examined (Scheme 3). Acetophenazine could react smoothly with various diazomalonates to afford the corresponding symmetrically substituted products in 86–94% yields. Either symmetrical or asymmetrical diazomalonates could be used as reactants, giving the symmetrical products **3ab–3ac** and asymmetrical products **3ad** in 86–94% yields. The aldazines **11–m** were also found to give the anticipated products **3lb** (81%), **3mb** (60%), and **3me** (53%).

The aldehyde and ketone carbonyl groups could be recovered by treatment of the azine products **3** with concentrated aqueous HCl. This was illustrated for the *o*-malonyl-acetophenazine primary product **3aa**, which gave the corresponding tricarbonyl product **5aa** in 75% yield (Scheme 4). α -Arylmalonates are useful intermediates in organic synthesis. As a possible synthetic application, the intramolecular Claisen/Dieckmann condensation of **5aa** or analogues could lead to functional naphthalene derivatives with biological or pharmaceutical properties, such as selective inhibition of the human leukocyte elastase.^{7b,14}

Scheme 2. Scope of the Azines^a



^aReaction conditions: 1 (0.2 mmol), 2a (0.45 mmol), $[Cp*RhCl_2]_2$ (2.5 mol %), AgOAc (10 mol %), 2,4,6-trimethylbenzoic acid (25 mol %), MeOH (1.5 mL), 12 h, under air, rt. Isolated yields. ^b60 °C.

Scheme 3. Scope of the Diazomalonates⁴



^aReaction conditions: 1 (0.2 mmol), 2 (0.45 mmol), [Cp*RhCl₂]₂ (2.5 mol %), AgOAc (10 mol %), 2,4,6-trimethylbenzoic acid (25 mol %), MeOH (1.5 mL), 12 h, under air, rt. Isolated yields. ^b60 °C.

CONCLUSION

In conclusion, we developed an efficient procedure for the regioselective Rh-catalyzed *ortho*-C-H carbenoid insertion

of α -diazomalonates with aryl-ketazine or -aldazine substrates. A wide range of *ortho*-substituted diarylazines were obtained in moderate to high yields at room temperature. The products Scheme 4. Example of Synthetic Application of the Rh-Catalyzed C–C Coupling of diaryl Azines with Dimethyl Diazomalonate to Access *o*-Malonyl-acetophenones



could be regarded as versatile intermediates to build valuable molecular targets.

EXPERIMENTAL SECTION

General Methods. All reactions were conducted within a tube under an air atmosphere. Unless otherwise stated, all commercial materials and solvents were used directly without further purification. ¹H and ¹³C NMR spectra were measured on a 400 MHz Bruker spectrometer (¹H 400 MHz, ¹³C 100 MHz), using CDCl₃ as the solvent and tetramethylsilane (TMS) as the internal standard at room temperature. Chemical shifts are reported in ppm referenced to the center of a triplet at 77.0 ppm of chloroform-*d*. High-resolution mass spectra (HRMS) were recorded using an Agilent 6450 spectrometer. High-resolution mass spectra (HRMS) were equipped with an ESI source and a TOF detector. Column chromatography was performed on silica gel (70–230 mesh ASTM) using the reported eluents. Thin-layer chromatography (TLC) was carried out on 4 cm × 15 cm plates with a layer thickness of 0.2 mm (silica gel 60 F254).

General Procedure for the Synthesis of Aromatic Ketazines 1 According to a Previously Reported Method. ¹⁵ Hydrazine hydrate (1.0 g, 16.6 mmol, 80%) was added to a solution of acetophenone (2.0 g, 16.6 mmol) and AcOH (119.0 mg, 1.7 mmol) in EtOH (15 mL), and then the mixture was heated to 100 °C with stirring and refluxed for 10 h. After cooling to room temperature, the mixture was filtered, washed with EtOH (40 mL × 4), and dried under vacuum. The corresponding bright yellow (1*E*,2*E*)-1,2-bis(1-phenylethylidene)hydrazine (1a) was obtained in 90% yield.

General Procedure for the Synthesis of Diazo Compounds 2 According to a Previously Reported Method.¹⁶ To a solution of dimethyl malonate (10 mmol) and *p*-acetamidobenzenesulfonyl azide (*p*-ABSA) (12 mmol) in anhydrous CH_3CN (20 mL) at 0 °C was added 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) (14 mmol) dropwise for 30 min. After stirring at room temperature overnight, the reaction mixture was concentrated under vacuum. Water (20 mL) was added. The resulting mixture was extracted with dichloromethane (2 × 20 mL). The combined organic layer was washed with brine (20 mL) and dried over MgSO₄. The solvent was removed under reduced pressure, and the residue was purified by silica gel column chromatography with petroleum ether/ethyl acetate as the eluent to give the title diazo compounds.

Preparation of Products 3. Under an air atmosphere, a reaction tube (5 mL) equipped with a magnetic stirring bar was charged with arylazine 1 (0.2 mmol), diazomalonate 2 (0.45 mmol), $[Cp*RhCl_2]_2$ (2.5 mol %), AgOAc (0.02 mmol, 10 mol %), 2,4,6-trimethylbenzoic acid (0.05 mmol, 25 mol %), and MeOH (1.5 mL). The reaction mixture was stirred at room temperature for 12 h. The solvents were removed under reduced pressure, and the residue was purified by chromatography on silica gel (eluent: petroleum ether/ethyl acetate = 4:1) to give the desired product.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(2,1-phenylene))dimalonate (**3aa**). Eluent: petroleum ether/ethyl acetate (4:1). White solid, 97% yield (96 mg), mp: 181–183 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.53–7.41 (m, 8H), 5.54 (s, 2H), 3.76 (s, 12H), 2.25 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.3, 161.2, 138.8, 130.7, 129.8, 129.1, 128.4, 128.1, 54.6, 52.7, 18.3. HRMS (ESI) *m/z*: calcd for C₂₆H₂₉N₂O₈⁺ (M + H⁺), 497.1918; found, 497.1926. Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(3-methyl-6,1-phenylene))dimalonate (**3ba**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 94% yield (99 mg), mp: 180–181 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.37 (d, *J* = 7.6 Hz, 2H), 7.29 (s, 2H), 7.22 (d, *J* = 8.0 Hz, 2H), 5.58 (s, 2H), 3.76 (s, 12H), 2.40 (s, 6H), 2.23 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.4, 161.0, 139.2, 136.0, 130.6, 130.3, 128.9, 128.4, 54.6, 52.7, 21.4, 18.2. HRMS (ESI) *m/z*: calcd for C₂₈H₃₃N₂O₈⁺ (M + H⁺), 525.2231; found, 525.2235.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(3-methoxy-6,1-phenylene))dimalonate (**3**ca). Eluent: petroleum ether/ethyl acetate (4:1). White solid, 92% yield (102 mg), mp: 177–178 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.44 (d, *J* = 8.4 Hz, 2H), 7.04 (d, *J* = 2.4 Hz, 2H), 6.93 (dd, *J* = 8.6, 2.5 Hz, 2H), 5.68 (s, 2H), 3.84 (s, 6H), 3.76 (s, 12H), 2.23 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.2, 160.8, 159.8, 132.5, 131.5, 130.0, 115.6, 113.3, 55.4, 54.7, 52.6, 18.1. HRMS (ESI) *m*/*z*: calcd for C₂₈H₃₃N₂O₁₀⁺ (M + H⁺), 557.2130; found, 557.2136.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(3-isobutyl-6,1-phenylene))dimalonate (**3da**). Eluent: petroleum ether/ethyl acetate (4:1). White solid, 74% yield (90 mg), mp: 140–141 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.40 (d, *J* = 7.6 Hz, 2H), 7.25 (d, *J* = 1.2 Hz, 2H), 7.19 (dd, *J* = 8.0, 1.3 Hz, 2H), 5.59 (s, 2H), 3.75 (s, 12H), 2.52 (d, *J* = 6.8 Hz, 4H), 2.25 (s, 6H), 1.87 (m, 2H), 0.92 (d, *J* = 6.8 Hz, 12H). ¹³C NMR (100 MHz, CDCl₃): δ 169.4, 161.1, 142.7, 136.2, 130.7, 130.4, 128.8, 128.2, 54.6, 52.6, 45.0, 30.1, 22.2, 18.2. HRMS (ESI) *m/z*: calcd for $C_{34}H_{45}N_2O_8^+$ (M + H⁺), 609.3170; found, 609.3178.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(3-(tert-butyl)-6,1-phenylene))dimalonate (**3ea**). Eluent: petroleum ether/ethyl acetate (4:1). White solid, 75% yield (91 mg), mp: 148–150 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.51 (s, 2H), 7.42 (s, 4H), 5.61 (s, 2H), 3.76 (s, 12H), 2.24 (s, 6H), 1.34 (s, 18H). ¹³C NMR (100 MHz, CDCl₃): δ 169.4, 161.0, 152.1, 136.0, 130.4, 128.2, 127.0, 125.0, 54.8, 52.6, 34.7, 31.1, 18.2. HRMS (ESI) *m/z*: calcd for $C_{34}H_{45}N_2O_8^+$ (M + H⁺), 609.3170; found, 609.3179.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(4-methyl-2,1-phenylene))dimalonate (**3fa**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 94% yield (99 mg), mp: 189–191 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.40 (d, *J* = 7.6 Hz, 2H), 7.26 (s, 2H), 7.23 (d, *J* = 8.0 Hz, 2H), 5.47 (s, 2H), 3.75 (s, 12H), 2.39 (s, 6H), 2.24 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.3, 161.2, 138.8, 137.9, 129.8, 129.6, 129.0, 127.7, 54.3, 52.6, 21.1, 18.3. HRMS (ESI) *m/z*: calcd for C₂₈H₃₃N₂O₈⁺ (M + H⁺), 525.2231; found, 525.2238.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(6-methoxy-2,1-phenylene))dimalonate (**3ga**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 63% yield (70 mg), mp: 173–175 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.38 (t, *J* = 8.0 Hz, 2H), 7.11–7.05 (m, 2H), 7.01 (d, *J* = 8.4 Hz, 2H), 5.32 (s, 2H), 3.84 (s, 6H), 3.73 (s, 12H), 2.21 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.1, 161.8, 158.4, 140.3, 128.9, 121.4, 120.7, 112.6, 56.3, 52.3, 51.5, 18.3. HRMS (ESI) *m/z*: calcd for C₂₈H₃₃N₂O₁₀⁺ (M + H⁺), 557.2130; found, 557.2138.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(6-fluoro-2,1-phenylene))dimalonate (**3ha**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 68% yield (72 mg), mp: 126–128 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.41 (m, 2H), 7.27 (d, J = 7.2 Hz, 2H), 7.20–7.14 (m, 2H), 5.43 (s, 2H), 3.77 (s, 12H), 2.24 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 168.3, 161.9 (d, J = 248.9 Hz),

The Journal of Organic Chemistry

161.1 (d, *J* = 2.5 Hz), 140.6 (d, *J* = 3.7 Hz), 129.5 (d, *J* = 9.5 Hz), 124.0 (d, *J* = 2.9 Hz), 120.2 (d, *J* = 15.1 Hz), 116.6 (d, *J* = 23.2 Hz), 52.7, 51.1, 18.2. ¹⁹F NMR (376 MHz, CDCl₃): δ –110.4. HRMS (ESI) *m/z*: calcd for C₂₆H₂₇F₂N₂O₈⁺ (M + H⁺), 533.1730; found, 533.1734.

Tetramethyl 2,2'-(((1E,1'É)-Hydrazine-1,2-diylidenebis(propan-1yl-1-ylidene))bis(2,1-phenylene))dimalonate (**3ia**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 85% yield (89 mg), mp: 153– 154 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.57 (d, *J* = 7.2 Hz, 2H), 7.46– 7.39 (m, 6H), 5.43 (s, 2H), 3.75 (s, 12H), 2.74 (d, *J* = 7.6 Hz, 4H), 0.99 (t, *J* = 7.5 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.1, 165.7, 138.0, 131.2, 129.9, 128.8, 127.9, 127.9, 54.2, 52.6, 24.9, 10.6. HRMS (ESI) *m/z*: calcd for C₂₈H₃₃N₂O₈⁺ (M + H⁺), 525.2231; found, 525.2238. Tetramethyl 2,2'-((8E,8'E)-Hydrazine-1,2-diylidenebis(5,6,7,8-tet-

Tetramethyl 2,2'-((8E,8'E)-Hydrazine-1,2-diylidenebis(5,6,7,8-tetrahydronaphthalene-1-yl-8-ylidene))dimalonate (**3***j*a). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 74% yield (81 mg), mp: 200–202 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.29 (m, 4H), 7.19 (d, *J* = 7.2 Hz, 2H), 6.01 (s, 2H), 3.75 (s, 12H), 2.79–2.74 (m, 4H), 2.58 (t, *J* = 6.6 Hz, 4H), 1.86–1.78 (m, 4H). ¹³C NMR (101 MHz, CDCl₃): δ 169.7, 156.9, 142.6, 131.9, 131.4, 128.9, 128.5, 128.3, 56.1, 52.5, 31.1, 28.8, 21.4. HRMS (ESI) *m/z*: calcd for C₃₀H₃₃N₂O₈⁺ (M + H⁺), 549.2231; found, 549.2238.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1yl-1-ylidene))bis(naphthalene-3,2-diyl))dimalonate (**3ka**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 74% yield (88 mg), mp: 229–230 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.99 (s, 2H), 7.95 (s, 2H), 7.90–7.85 (m, 4H), 7.56–7.52 (m, 4H), 5.77 (s, 2H), 3.81 (s, 12H), 2.37 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.5, 160.5, 136.2, 133.0, 132.3, 129.4, 128.7, 128.5, 128.0, 127.8, 127.2, 127.0, 55.1, 52.8, 18.2. HRMS (ESI) *m/z*: calcd for $C_{34}H_{33}N_2O_8^+$ (M + H⁺), 597.2231; found, 597.2236.

Tetramethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis-(methanylylidene))bis(3-methyl-2,1-phenylene))dimalonate (**3la**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 95% yield (94 mg), mp: 230–232 °C. ¹H NMR (400 MHz, CDCl₃): δ 8.97 (s, 2H), 7.36 (t, *J* = 7.6 Hz, 2H), 7.25 (d, *J* = 6.8 Hz, 4H), 5.71 (s, 2H), 3.78 (s, 12H), 2.50 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.3, 161.7, 139.5, 132.6, 130.9, 130.7, 130.5, 127.7, 56.1, 52.7, 20.3. HRMS (ESI) *m/z*: calcd for $C_{26}H_{29}N_2O_8^+$ (M + H⁺), 497.1918; found, 497.1926.

Tetramethyl² 2, 2'-(((1 E, 1' E)-Hydrazine-1, 2-diylidenebis-(methanylylidene))bis(4-methyl-2, 1-phenylene))dimalonate (**3ma**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 75% yield (74 mg), mp: 175–177 °C. ¹H NMR (400 MHz, CDCl₃): δ 8.73 (s, 2H), 7.56 (s, 2H), 7.34 (d, J = 8.0 Hz, 2H), 7.28 (d, J = 8.0 Hz, 2H), 5.80 (s, 2H), 3.78 (s, 12H), 2.40 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.0, 162.7, 138.3, 132.3, 131.9, 131.8, 129.9, 129.8, 54.6, 52.8, 20.9. HRMS (ESI) *m/z*: calcd for C₂₆H₂₉N₂O₈⁺ (M + H⁺), 497.1918; found, 497.1925.

Tetraethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1-yl-1-ylidene))bis(2,1-phenylene))dimalonate (**3ab**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 94% yield (104 mg), mp: 100–101 °C. ¹H NMR (400 MHz, CDCl₃): δ 7.55 (dd, *J* = 6.9, 1.8 Hz, 2H), 7.48–7.37 (m, 6H), 5.48 (s, 2H), 4.29–4.17 (m, 8H), 2.26 (s, 6H), 1.26 (t, *J* = 7.1 Hz, 12H). ¹³C NMR (100 MHz, CDCl₃): δ 168.7, 160.9, 139.0, 130.9, 129.8, 128.9, 128.2, 127.9, 61.5, 54.8, 18.5, 14.0. HRMS (ESI) *m/z*: calcd for $C_{30}H_{37}N_2O_8^+$ (M + H⁺), 553.2544; found, 553.2553.

Tetrabenzyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis(ethan-1-yl-1-ylidene))bis(2,1-phenylene))dimalonate (**3ac**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow oil, 87% yield (139 mg). ¹H NMR (400 MHz, CDCl₃): δ 7.47 (dd, J = 5.2, 2.4 Hz, 2H), 7.34 (d, J = 2.0 Hz, 6H), 7.23 (s, 20H), 5.65 (s, 2H), 5.14 (d, J = 2.4 Hz, 8H), 2.07 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 168.4, 160.9, 138.8, 135.3, 129.9, 128.9, 128.5, 128.4, 128.3, 128.2, 128.1, 128.0, 127.9, 127.8, 67.2, 54.8, 184. HRMS (ESI) *m*/*z*: calcd for C₅₀H₄₅N₂O₈⁺ (M + H⁺), 801.3170; found, 801.3175.

1-Benzyl-3-methyl-2-(2-((E)-1-(((E)-1-(2-(1-(benzyloxy)-3-methoxy-1,3-dioxopropan-2-yl)phenyl)ethylidene)hydrazono)ethyl)-phenyl)malonate (**3ad**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow oil, 86% yield (111 mg). ¹H NMR (400 MHz, CDCl₃): δ 7.48 (d, *J* = 2.8 Hz, 2H), 7.43–7.36 (m, 6H), 7.28 (s, 10H), 5.59 (d, *J* = 2.0 Hz, 2H), 5.19 (d, *J* = 5.2 Hz, 4H), 3.70 (d, *J* = 5.2 Hz, 6H), 2.16

(s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 169.1, 168.5, 161.0, 138.8, 135.4, 130.6, 129.9, 129.0, 128.4, 128.3, 128.2, 128.0, 128.0, 67.2, 54.7, 52.6, 18.4. HRMS (ESI) *m*/*z*: calcd for C₃₈H₃₇N₂O₈⁺ (M + H⁺), 649.2544; found, 649.2549.

Tetraethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis-(methanylylidene))bis(3-methyl-2,1-phenylene))dimalonate (**3** *lb*). Eluent: petroleum ether/ethyl acetate (4:1). White solid, 81% yield (89 mg), mp: 108–110 °C. ¹H NMR (400 MHz, CDCl₃): δ 8.98 (s, 2H), 7.36 (t, *J* = 7.6 Hz, 2H), 7.29 (d, *J* = 7.6 Hz, 2H), 7.24 (d, *J* = 7.6 Hz, 2H), 5.68 (s, 2H), 4.25 (m, 8H), 2.49 (s, 6H), 1.28 (t, *J* = 7.1 Hz, 12H). ¹³C NMR (100 MHz, CDCl₃): δ 168.9, 161.7, 139.4, 132.9, 131.0, 130.6, 130.3, 127.7, 61.6, 56.2, 20.4, 14.1. HRMS (ESI) *m/z*: calcd for C₃₀H₃₇N₂O₈⁺ (M + H⁺), 553.2544; found, 553.2538.

Tetraethyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis-(methanylylidene))bis(4-methyl-2,1-phenylene))dimalonate (**3mb**). Eluent: petroleum ether/ethyl acetate (4:1). White solid, 60% yield (66 mg), mp: 145–146 °C. ¹H NMR (400 MHz, CDCl₃): δ 8.77 (s, 2H), 7.58 (s, 2H), 7.37 (d, *J* = 8.0 Hz, 2H), 7.28 (d, *J* = 8.0 Hz, 2H), 5.72 (s, 2H), 4.27–4.23 (m, 6H), 2.40 (s, 6H), 1.27 (t, *J* = 7.1 Hz, 12H). ¹³C NMR (100 MHz, CDCl₃): δ 168.6, 162.5, 138.1, 132.0, 131.7, 130.1, 129.9, 128.6, 61.7, 54.8, 20.9, 14.0. HRMS (ESI) *m/z*: calcd for C₃₀H₃₇N₂O₈⁺ (M + H⁺), 553.2544; found, 553.2540.

Tetraisopropyl 2,2'-(((1E,1'E)-Hydrazine-1,2-diylidenebis-(methanylylidene))bis(4-methyl-2,1-phenylene))dimalonate (**3me**). Eluent: petroleum ether/ethyl acetate (4:1). Yellow solid, 53% yield (64 mg), mp: 143–145 °C. ¹H NMR (400 MHz, CDCl₃): δ 8.78 (s, 2H), 7.59 (s, 2H), 7.38 (d, *J* = 8.0 Hz, 2H), 7.27 (d, *J* = 9.2 Hz, 2H), 5.62 (s, 2H), 5.10 (m, 4H), 2.40 (s, 6H), 1.28 (d, *J* = 6.0 Hz, 12H), 1.25 (d, *J* = 6.4 Hz, 12H). ¹³C NMR (100 MHz, CDCl₃): δ 168.1, 162.3, 137.9, 132.0, 131.6, 130.4, 129.9, 128.6, 69.2, 55.0, 21.59, 21.58, 20.9. HRMS (ESI) *m/z*: calcd for C₃₄H₄₅N₂O₈⁺ (M + H⁺), 609.3170; found, 609.3171.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.7b01472.

Description of the X-ray structures and NMR spectra of the compounds (PDF)

Crystal data of the compounds (CIF)

AUTHOR INFORMATION

Corresponding Authors

*E-mail: remi.chauvin@lcc-toulouse.fr.

*E-mail: cuixl@hqu.edu.cn. Fax: (+)86-592-6162996.

ORCID 0

Remi Chauvin: 0000-0002-4491-6390

Xiuling Cui: 0000-0001-5759-766X

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This research was supported by NSF of China (21572072), Xiamen Southern Oceanographic Center (15PYY052SF01), Science and Technology Bureau of Xiamen City (3502Z20150054), and Huaqiao University.

REFERENCES

(1) For recent reviews of catalytic C-H functionalizations, see: (a) Wencel-Delord, J.; Dröge, T.; Liu, F.; Glorius, F. Chem. Soc. Rev. **2011**, 40, 4740. (b) Ackermann, L.; Vicente, R.; Kapdi, A. R. Angew. Chem., Int. Ed. **2009**, 48, 9792. (c) Yeung, C. S.; Dong, V. M. Chem. Rev. **2011**, 111, 1215. (d) Colby, D. A.; Tsai, A. S.; Bergman, R. G.; Ellman, J. A. Acc. Chem. Res. **2012**, 45, 814. (e) Song, G.; Wang, F.; Li, X. Chem. Soc. Rev. **2012**, 41, 3651. (f) Satoh, T.; Miura, M. Chem. - Eur. J. **2010**, 16,

The Journal of Organic Chemistry

11212. (g) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Chem. Rev. 2010, 110, 624.

(2) (a) Lyons, T. W.; Sanford, M. S. Chem. Rev. 2010, 110, 1147.
(b) Daugulis, O. Top. Curr. Chem. 2009, 292, 57. (c) Chen, X.; Engle, K. M.; Wang, D.-H.; Yu, J.-Q. Angew. Chem., Int. Ed. 2009, 48, 5094.
(d) Beccalli, E. M.; Broggini, G.; Martinelli, M.; Sottocornola, S. Chem. Rev. 2007, 107, 5318. (e) Ritleng, V.; Sirlin, C.; Pfeffer, M. Chem. Rev. 2002, 102, 1731.

(3) (a) Ryabov, A. D. Synthesis 1985, 1985, 233. (b) Crabtree, R. H. Chem. Rev. 1985, 85, 245. (c) Ryabov, A. D. Chem. Rev. 1990, 90, 403. (d) Shilov, A. E.; Shul'pin, G. B. Chem. Rev. 1997, 97, 2879. (e) Zhang, X. S.; Chen, K.; Shi, Z. J. Chem. Sci. 2014, 5, 2146. (f) Zhang, M.; Zhang, Y.; Jie, X.; Zhao, H.; Li, G.; Su, W. Org. Chem. Front. 2014, 1, 843. (g) Chen, Z.; Wang, B.; Zhang, J.; Yu, W.; Liu, Z.; Zhang, Y. Org. Chem. Front. 2015, 2, 1107. (h) Shan, G.; Huang, G.-Y.; Rao, Y.; Zhang, H. Chin. Chem. Lett. 2015, 26, 1236.

(4) (a) Yu, S.; Li, X. Org. Lett. **2014**, *16*, 1220. (b) Zhang, G.; Yang, L.; Wang, Y.; Xie, Y.; Huang, H. J. Am. Chem. Soc. **2013**, *135*, 8850. (c) Ali, M. A.; Yao, X.; Li, G.; Lu, H. Org. Lett. **2016**, *18*, 1386. (d) Qu, S.; Cramer, C. J. J. Org. Chem. **2017**, *82*, 1195.

(5) Yu, X.; Yu, S.; Xiao, J.; Wan, B.; Li, X. J. Org. Chem. 2013, 78, 5444.
(6) (a) Ueura, K.; Satoh, T.; Miura, M. Org. Lett. 2005, 7, 2229.
(b) Fukutani, T.; Umeda, N.; Hirano, K.; Satoh, T.; Miura, M. Chem. Commun. 2009, 34, 5141. (c) Zhang, T.; Wu, L.; Li, X. Org. Lett. 2013, 15, 6294. (d) Wang, N. J.; Mei, S. T.; Shuai, L.; Yuan, Y.; Wei, Y. Org. Lett. 2014, 16, 3040.

(7) (a) Tsai, A. S.; Brasse, M.; Bergman, R. G.; Ellman, J. A. Org. Lett. 2011, 13, 540. (b) Chan, W.-W.; Lo, S.-F.; Zhou, Z.; Yu, W.-Y. J. Am. Chem. Soc. 2012, 134, 13565. (c) Ng, K.-H.; Zhou, Z.; Yu, W.-Y. Chem. Commun. 2013, 49, 7031.

(8) (a) Chuang, S.-C.; Gandeepan, P.; Cheng, C.-H. Org. Lett. 2013, 15, 5750. (b) Huang, X.-C.; Yang, X.-H.; Song, R.-J.; Li, J.-H. J. Org. Chem. 2014, 79, 1025.

(9) (a) Tang, W.; Xiang, Y.; Tong, A. J. Org. Chem. 2009, 74, 2163.
(b) Le Goff, G.; Ouazzani, J. Bioorg. Med. Chem. 2014, 22, 6529.

(c) Blair, L. M.; Sperry, J. J. Nat. Prod. 2013, 76, 794.

(10) (a) Manikannan, R.; Venkatesan, R.; Muthusubramanian, S.; Yogeeswari, P.; Sriram, D. Bioorg. Med. Chem. Lett. 2010, 20, 6920.
(b) Huang, X. C.; Yang, X. H.; Song, R. J.; Li, J. H. J. Org. Chem. 2014, 79, 1025.
(c) Cohen, R.; Rybtchinski, B.; Gandelman, M.; Shimon, L. J.; Martin, J. M.; Milstein, D. Angew. Chem., Int. Ed. 2003, 42, 1949.
(d) Han, W.; Zhang, G.; Li, G.; Huang, H. Org. Lett. 2014, 16, 3532.

(11) Han, W.; Zhang, G.; Li, G.; Huang, H. Org. Lett. **2014**, *16*, 3532. (12) Wen, J.; Wu, A.; Wang, M.; Zhu, J. J. Org. Chem. **2015**, *80*, 10457.

(13) (a) Zhang, M.; Zhang, Y.; Jie, X.; Zhao, H.; Li, G.; Su, W. Org. Chem. Front. **2014**, *1*, 843. (b) Zhang, T.; Wu, L.; Li, X. Org. Lett. **2013**, *15*, 6294. (c) Wen, J.; Wu, A.; Wang, M.; Zhu, J. J. Org. Chem. **2015**, *80*, 10457. (d) Zhang, Y.-F.; Wu, B.; Shi, Z.-J. Chem. - Eur. J. **2016**, *22*, 17808. (e) Li, Y.; Feng, Y.; Xu, L.; Wang, L.; Cui, X. Org. Lett. **2016**, *18*, 4924. (f) Xia, Y.; Liu, Z.; Feng, S.; Zhang, Y.; Wang, J. J. Org. Chem. **2015**, *80*, 223.

(14) (a) Elsebai, M. F.; Natesan, L.; Kehraus, S.; Mohamed, I. E.; Schnakenburg, G.; Sasse, F.; Shaaban, S.; Gütschow, M.; König, G. M. J. *Nat. Prod.* **2011**, *74*, 2282. (b) de Koning, C. B.; Rousseau, A. L.; van Otterlo, W. A. L. *Tetrahedron* **2003**, *59*, 7.

(15) (a) Daub, G. H.; Cannizzo, L. F. J. Org. Chem. 1982, 47, 5034.
(b) Han, W.; Zhang, G.; Li, G.; Huang, H. Org. Lett. 2014, 16, 3532.

(16) Koduri, N. D.; Scott, H.; Hileman, B.; Cox, J. D.; Coffin, M.; Glicksberg, L.; Hussaini, S. R. *Org. Lett.* **2012**, *14*, 440.